Types of Processes

A fixed mass (simple compressible substance) system can undergo different types of processes between state 1 and state 2: Some types are given below:

- Isothermal process constant temperature T = C
- Isobaric process constant pressure P = C
- Isochoric process constant volume V = C
- Adiabatic process no heat transfer Q = 0
- Polytropic process $PV^n = C$, $0 < n < \infty$
- Quasistatic process consists of succession of equilibrium states
- Reversible process initial state of the system can be restored
- Irreversible process initial state of the system cannot be restored without observable effects in the system and its surroundings